# **Atomic Energy Central School No 4 Rawatbhata**

# Multiple Choice Questions Examination (November 2019-20)

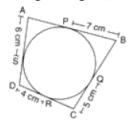
MM: 120 Class X (Mathematics, Science, Social Science) Time:3hour

Name of student : _	Roll No	Class Sec

Date: \_\_\_\_\_\_ Invigilator's Sign: \_\_\_\_\_

## **Mathematics**

1. In the given figure, the perimeter of ABCD is



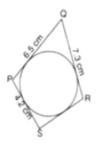
a) 44 cm

b) 36 cm

c) 40 cm

d) 48 cm

2. In the given figure, a circle touches all four sides of a quadrilateral PQRS, whose sides are PQ = 6.5 cm, QR = 7.3 cm, and PS = 4.2 cm, then RS is equal to



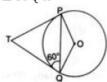
a) 7.3 cm

b) 5.3 cm

c) 4.7 cm

d) 5 cm

3. TP and TQ are tangents from an external point T, to a circle with centre O ∠TPQ = 60° then the measure of ∠OPQ is



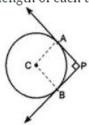
a) 60°

b) 90°

·) 40°

d) 30°

4. In the given figure, the pair of tangents A to a circle with centre O are perpendicular to each other and length of each tangent is 5 cm, then the radius of the circle is :



a) 2.5 cm

b) 5 cm

c) 7.5 cm

d) 10 cm

5. Two concentric circles of radii 3 cm and 5 cm are given. The length of chord BC which touches the inner circle at P is equal to

1

1



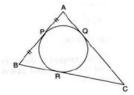
b) 8 cm

c) 6 cm.

d) 10 cm.

6. In the given figure, if AP = PB, then AC =

1



a) AC = BC

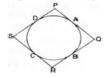
b) AB = BC

c) AQ = QC

d) AC = AB

7. Quadrilateral PQRS circumscribes a circle as shown in the figure. The side of the quadrilateral which is equal to PD + QB is





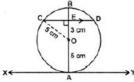
a) PS

b) PR

c) QR

d) PQ

8. At one end A of a diameter AB of a circle of radius 5 cm, tangent XAY is drawn to the circle. The length of the chord CD parallel to XY and at a distance 8 cm from A is:



a) 5 cm

b) 6 cm

c) 8 cm

d) 4 cm

9. The given figure shows two concentric circles with centre O. PR and PQS are tangents to the inner circle from point P lying on the outer circle. If PR = 7.5 cm, then PS is equal to





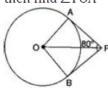
a) 10 cm.

b) 18 cm.

c) 12 cm.

d) 15 cm.

10. If tangents PA and PB from a point P to a circle with centre O are inclined to each other at an angle of 80°, 1 then find ∠POA

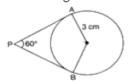


a) 40°

b) 100°

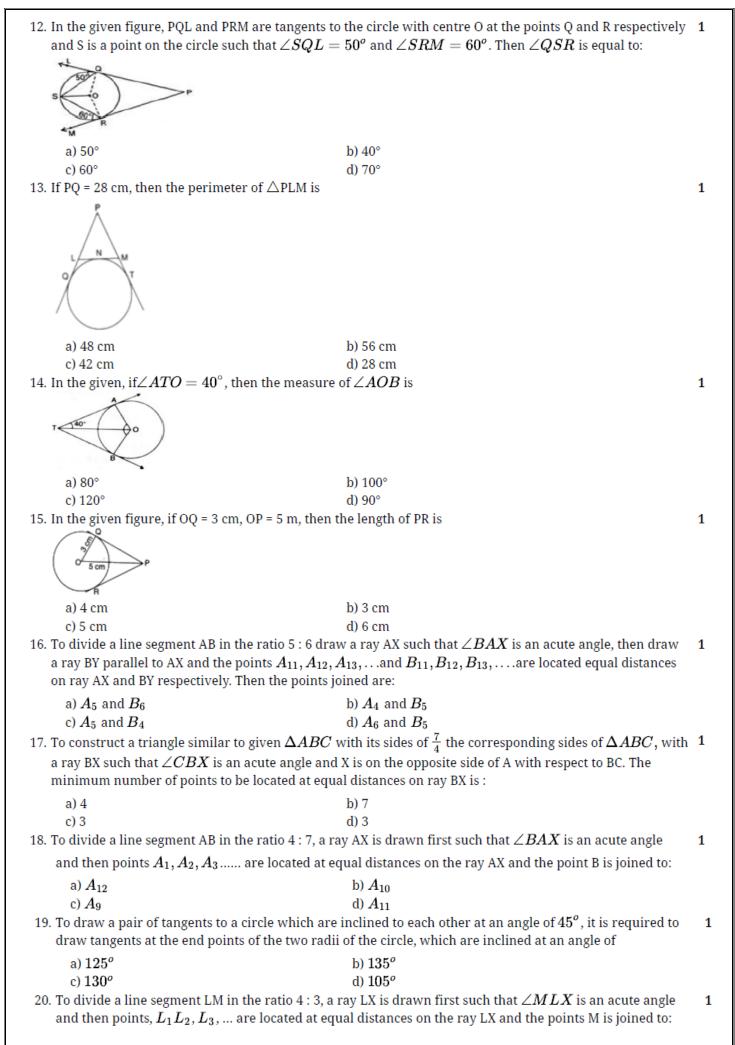
d) 60°

11. If two tangents inclined at 60° are drawn to circle of radius 3 cm, then length of each tangent is equal to 1



b) 3 cm

d)  $3\sqrt{2}$  cm



a) $L_2$	b) $L_7$	
c) $L_4$	d) $L_3$	
21. If the point R(-2,3) divides the line segment PQ in	the ratio 3:4 where the coordinates of P are (-6,7), then	1
the coordinates of Q are:		
a) $(\frac{10}{3}, \frac{7}{3})$	b) (7,8)	
c) (4,5)	d) (3,8)	
	w a ray AX such that is $\angle BAX$ an acute angle, then draw	1
a ray BT $\mid \mid$ AX and mark the points and $A_1, A_2, A_3$	· ·	
	3 = on the rays AX and BY respectively. Then, the points	
to be joined are :		
a) $A_5 \ to \ B_2$	b) $A_7$ to $B_5$	
c) $A_2$ to $B_5$	d) $A_5$ to $B_7$	
23. To divide a line segment AB in the ration 5: 6, dra	aw a ray AX such that is $\angle BAX$ an acute angle, then draw	1
a ray BY parallel to AX and the points $A_1,A_2,A_3$	$a\ldots$ and are $B_1,B_2,B_3\ldots$ located at equal distances on ray	
AX and BY, respectively. Then, the points joined a	are	
a) $A_5$ $and$ $B_4$	b) $A_6$ and $B_5$	
c) $A_4$ and $B_5$	d) $A_5$ and $B_6$	
24. To divide a line segment AB in the ratio p: $\boldsymbol{q}$ ( $\boldsymbol{p},$	${f q}$ are positive integers), draw a ray AX so that $\angle BAX$ is an	1
acute angle and then mark points on ray AX at e	qual distances such that the minimum number of these	
points is:		
a) pq	b) p + q	
c) p + q – 1	d) greater of p and q	
25. In the given figure, $AA_1 = A_1A_2 = A_2A_3 = A_3C$		1
IF $B_1A_1 \mid \mid$ CB, then $A_1$ divides AC in the ratio		
B <sub>2</sub> B		
B <sub>1</sub> B <sub>2</sub>		
A A1 A2 A2 C		
a) 4:1	b) 1:3	
c) 1/4	d) 1:2	
26. If a line meets the circle in two distinct points, it		1
a) a chord	b) a radius	_
c) secant	d) a tangent	
27. The approximate value of $\pi = \frac{62832}{20000}$ is given by		1
		•
a) Bhaskara	b) Ramanujan	
c) Aryabhatta	d) Mahveera	
28. If the area of a circle is equal to the area of a squ	are, then the ratio of their perimeters is	1
a) 2 : π	b) 1:2	
c) $\sqrt{\pi}:2$	$d) \pi : 2$	
29. The area of a circle is 2464 sq. cm, then its diame		1
a) 28cm	b) 7cm	
c) 14cm	d) 56cm	
30. The area of a ring having 'R' as outer radius and	'r' as inner radius is	1
a) $\pi(R+r)$	b) $\pi(R^2-r^2)$	
c) $\pi(R^2+r^2)$	d) $\pi(R-r)$	
31. If the sum of the areas of two circles with radii r	$_{1}$ and $_{2}$ is equal to the area of the circle of radius R, then	1
a) $R^2 = {r_1}^2 - {r_2}^2$	b) p3- n 2 · n 2	
	b) $R^3 = r_1^2 + r_2^2$	
c) $R^2 + {r_1}^2 = {r_2}^2$	d) $R^2 = r_1^2 + r_2^2$	
32. The diameter of the circle whose area is $301.84$	cm² is	1
a) 14.2 cm	b) 12.8 cm	
c) 19.6 cm	d) 15.6 cm	

<ol> <li>If the sum of the circumfered circle of radius R, then</li> </ol>	ences of two circles	with radii $R_1 and \ R_2$ is equal to the circumference of a	1
a) $R_1 + R_2 < R$		b) $R_1-\ R_2=\ R$	
c) $R_1 + R_2 > R$		d) $R_1 + R_2 = R$	
34. If the perimeter of a circle i	is equal to that of a s	quare, then the ratio of their areas is	1
a) 22:7		b) 14:11	
c) 11:14		d) 7:22	
-	f a circle are numeri	ically equal, then the radius of the circle is	1
a) 7 units.		b) 2 units	
c) 4 units		d) 5 units	
	ice of a circle are nu	merically equal, then its radius is	1
a) π units c) 2π units		b) 2 units d) 4 units	
	ose circumference is	equal to the sum of the circumferences of the two circles of	1
radii 24cm and 7cm is		or and the carrier and constant control of the two controls of	-
a) 25cm		b) 7cm	
c) 31cm		d) 24cm	
38. The perimeter of a protract	or is		1
a) $\pi r$		b) $\pi r  +  2r$	
c) $\pi + r$		d) $\pi + 2r$	
39. The perimeter of a semiciro	cular protractor who	ose radius is 7cm is	1
a) 18cm		b) 27cm	
c) 72cm	. h	d) 36cm	
The radius of the circle is	s bent into the form	of an arc of a circle subtending an angle of 60 at its centre.	1
a) $\frac{20}{6+\pi}$ cm		b) $\frac{30}{6+\pi}$ cm	
c) $\frac{60}{6+\pi}$ cm		d) $\frac{15}{6+\pi}$ cm	
		Science	
41. Butanone is a four carbon of	compound with the	functional group:	1
a) - COOH	compound with the	b) - CHO	-
c) - CO -		d) - OH	
42. Statement A: Valerie acid is	the common name	of hexane.	1
Statement B: Glycerol is add	led in the manufacti	uring of soap.	
a) Statement A is true and false	statement B is	b) Both the statements A and B are false	
c) Statement B is true and	statement A is	d) Neither statement A nor statement B is	
false		false	
43. Match the following with th		_	1
(1) Welding	(A) Ethyl alcohol	_	
(2) A fuel in spirit lamp	(B) Ester	_	
(3) Lime water turns milky	(C) Carbon dioxide	_	
(4) Fruity smell	(D) Ethyne and oxy	gen	
a) 1-D, 2-A, 3-C, 4-B		b) 1-A, 2-C, 3-B, 4-D	
c) 1-C, 2-B, 3-D, 4-A		d) 1-B, 2-D, 3-A, 4-C	4
44. What is denatured spirit?  a) None of these		h) Ethanol only	1
a) None of these c) Methanol only		b) Ethanol only d) Ethanol + Methanol (5%)	
45. Hydrocarbon with molecul	ar formula 4H <sub>10</sub> ha:		1
a) 10 covalent bonds		b) 7 covalent bonds	
c) 13 covalent bonds		d) 6 covalent bonds	
		5	

a) C <sub>3</sub> H <sub>6</sub> , C <sub>2</sub> H <sub>2</sub> b) C <sub>2</sub> H <sub>8</sub> d) C <sub>3</sub> H <sub>8</sub> c) C <sub>2</sub> H <sub>8</sub> C) C <sub>3</sub> H <sub>8</sub> c) C <sub>4</sub> H <sub>8</sub> C) C <sub>4</sub> H <sub>8</sub> c) C <sub>2</sub> H <sub>8</sub> C) Coal d) Craphite c) Coal d) Diamond d.	46. Which of the following hy	ydrocarbons undergo	addition reactions? $3H_8, C_3H_6, C_2H_5OH, CH_4, C_2H_2$	1
a) Fullerenes c) Coal d) Diamond  48. Drinking alcohol and driving may cause serious accidents. To discourage this, police randomly lest drivers 1 for alcohol using a breath analyzer. The breath analyzer works because: a) Alcohol makes the breath dry and the machine registers moisture. c) Alcohol causes more saliva which the machine checks.  49. The formulae of two organic acids 'X' and 'Y' are 10H21COOH and C19H39COOH. Which of them 1 exists in liquid state at room temperature? a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl2 d) CaSO4  51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false. 52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent d) Oxidizing agent e) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound 54. Match the following with the correct response: (1) Addition reaction (2) Substitution reaction (B) Methanol	c) $\mathrm{C_2H_5OH}, \mathrm{C_3H_8}$		d) $C_3H_8$ , $C_3H_6$	
d) Diamond  48. Drinking alcohol and driving may cause serious accidents. To discourage this, police randomly lest drivers 1 for alcohol using a breath analyzer. The breath analyzer works because:  a) Alcohol makes the breath hotter which changes the machine registers moisture. c) Alcohol causes more saliva which the machine checks. d) Alcohol in the breath cause a chemical change which is registered by the breath analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21COOH and C19H39COOH. Which of them  exists in liquid state at room temperature? a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl2 d) CaSO <sub>4</sub> c) CaCl2 d) CaSO <sub>4</sub> Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent d) Oxidizing agent e) Dybutton of a pungent smelling gas. d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (2) Substitution reaction (3) Methanol	•	members of which ty	-	1
48. Drinking alcohol and driving may cause serious accidents. To discourage this, police randomly lest drivers 1 for alcohol using a breath analyzer. The breath analyzer works because:  a) Alcohol makes the breath dry and the machine registers moisture. c) Alcohol causes more saliva which the machine checks.  49. The formulae of two organic acids 'X' and 'Y' are 10H21 COOH and C19H39COOH. Which of them analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21 COOH and C19H39COOH. Which of them 1 exists in liquid state at room temperature? a) Neither X and Y b) Both X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these b) Mg(HCO3)2 c) CaCl2 c) d) CaSO4 51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement A is false. 52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent b) Reducing agent c) Dehydrating agent d) Oxidizing agent o) Oxidizing agent o) Oxidizing agent sould be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound 54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol	,			
for alcohol using a breath analyzer. The breath analyzer works because:  a) Alcohol makes the breath dry and the machine registers moisture. c) Alcohol causes more saliva which the machine checks. d) Alcohol in the breath cause a chemical change which is registered by the breath analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21 COOH and C19H39COOH. Which of them  exists in liquid state at room temperature? a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl2 d) CaSO4  51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false. 52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent c) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (2) Substitution reaction (B) Methanol	•			1
machine registers moisture. c) Alcohol causes more saliva which the machine checks. d) Alcohol in the breath cause a chemical change which is registered by the breath analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21 COOH and C19H39COOH. Which of them 1  exists in liquid state at room temperature? a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl2 d) CaSO4  51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol	for alcohol using a breath	analyzer. The breath	analyzer works because:	rs 1
c) Alcohol causes more saliva which the machine checks.  change which is registered by the breath analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21COOH and C19H39COOH. Which of them 1  exists in liquid state at room temperature?  a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl2 d) CaSO4  51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true; c) Statement B is true; Statement A is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent e) Dehydrating agent d) Oxidizing agent e) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol		•		
machine checks.  change which is registered by the breath analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21 COOH and C19 H39 COOH. Which of them  exists in liquid state at room temperature?  a) Neither X and Y c) Y b) Both X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl2 d) CaSO4  51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false. 52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent swould be observed? a) Evolution of brown fumes c) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol	9		3	
analyzer machine.  49. The formulae of two organic acids 'X' and 'Y' are 10H21COOH and C19H39COOH. Which of them  exists in liquid state at room temperature?  a) Neither X and Y b) Both X and Y c) Y d) X  50. Hardness of water is caused by:  a) All of these b) Mg(HCO3)2 c) CaCl2 d) CaSO4  51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl2 to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent d) Oxidizing agent swould be observed? a) Evolution of brown fumes c) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response: (1) Addition reaction (3) Hydrogenation (2) Substitution reaction (6) Methanol		anva which the		
exists in liquid state at room temperature?  a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these c) CaCl <sub>2</sub> d) CaSO <sub>4</sub> 51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent solution of brown fumes would be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response: (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol				
a) Neither X and Y c) Y d) X  50. Hardness of water is caused by: a) All of these b) Mg(HCO <sub>3</sub> ) <sub>2</sub> c) CaCl <sub>2</sub> d) CaSO <sub>4</sub> 51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent solution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response: (1) Addition reaction (2) Substitution reaction (B) Methanol	49. The formulae of two orga	nic acids 'X' and 'Y' a	re $10 ext{H}_{21} ext{COOH}$ and $ ext{C}_{19} ext{H}_{39} ext{COOH}$ . Which of them	1
c) Y  50. Hardness of water is caused by:  a) All of these b) Mg(HCO <sub>3</sub> ) <sub>2</sub> c) CaCl <sub>2</sub> d) CaSO <sub>4</sub> 51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride.  a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent swould be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol	exists in liquid state at roo	om temperature?		
30. Hardness of water is caused by:  a) All of these c) CaCl <sub>2</sub> d) CaSO <sub>4</sub> 51. Statement A: Oxyacetylene flame is used for welding purposes. Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride.  a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent swould be observed? a) Evolution of brown fumes c) Brisk effervescence d) Production of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol				
a) All of these c) CaCl <sub>2</sub> d) CaSO <sub>4</sub> 51. Statement A: Oxyacetylene flame is used for welding purposes.  Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride. a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent swould be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol	· ·		d) X	
c) CaCl <sub>2</sub> d) CaSO <sub>4</sub> 51. Statement A: Oxyacetylene flame is used for welding purposes.  Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride.  a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent 53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol		ed by:	LVM (TIGO.)	1
Statement A: Oxyacetylene flame is used for welding purposes.  Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride.  a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent swould be observed? a) Evolution of brown fumes c) Brisk effervescence d) Production of a pungent smelling gas. d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (3) Methanol	•			
Statement B: Ethyne reacts with HCl in the presence of HgCl <sub>2</sub> to form vinyl chloride.  a) Both the statements A and B are true. b) Neither statement A nor statement B is true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent c) Dehydrating agent d) Oxidizing agent 53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (3) Substitution reaction (4) Hydrogenation (2) Substitution reaction (B) Methanol		e flame is used for we	•	1
true. c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a: a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent  53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed? a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol				-
c) Statement B is true; Statement A is false. d) Statement A is true; Statement B is false.  52. Activated charcoal is used in sugar industry as a:  a) Decolorizing agent b) Reducing agent c) Dehydrating agent d) Oxidizing agent  53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed?  a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol	a) Both the statements A	A and B are true.	b) Neither statement A nor statement B is	
52. Activated charcoal is used in sugar industry as a:  a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent 53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed?  a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol				
a) Decolorizing agent c) Dehydrating agent d) Oxidizing agent 53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed? a) Evolution of brown fumes c) Brisk effervescence b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound 54. Match the following with the correct response: 1 (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol				4
c) Dehydrating agent d) Oxidizing agent  53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed?  a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol		i in sugar industry as a		1
53. A few drops of ethanoic acid are added to solid sodium bicarbonate. Which of the following observations would be observed?  a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  1  (1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol				
a) Evolution of brown fumes b) Evolution of a pungent smelling gas. c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (2) Substitution reaction (B) Methanol	, ,	cid are added to solid		1
c) Brisk effervescence d) Production of a hissing sound  54. Match the following with the correct response:  (1) Addition reaction (A) Hydrogenation  (2) Substitution reaction (B) Methanol	would be observed?			
54. Match the following with the correct response:  (1) Addition reaction (2) Substitution reaction (B) Methanol	-	umes		
(1) Addition reaction (A) Hydrogenation (2) Substitution reaction (B) Methanol		the correct response:	d) Production of a hissing sound	1
(2) Substitution reaction (B) Methanol			$\neg$	1
		, , ,	$\dashv$	
1(2) Departuration 1(C) Corpor totrochloride	(3) Denaturation	(C) Carbon tetrachlor		
	**		ide	
(4) Esterification (D) Ethyl ethanoate		(ט) Etnyl ethanoate		
a) 1-B, 2-D, 3-A, 4-C b) 1-D, 2-A, 3-C, 4-B c) 1-C, 2-B, 3-D, 4-A d) 1-A, 2-C, 3-B, 4-D				
55. Statement A: Ethane decolorizes bromine water whereas ethyne does not.				1
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.	a) Statement B is true; S	statement A is false.	b) Both - Statement A and Statement B - are	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.			true.	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true.	c) Statement A is true; S	statement B is false.		
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are	56. Disposable plastic plates s	should not be used bec		1
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.	a) They are made of no	n-biodegradable	b) They are made of biodegradable	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because:  1	material.		materials.	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because: 1 a) They are made of non-biodegradable material. b) They are made of biodegradable materials.	c) They are made of tox	ic materials.	_	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because: 1 a) They are made of non-biodegradable material. b) They are made of biodegradable materials. c) They are made of toxic materials. d) They are made of materials with light			weight.	
		lantara b		4
55. Statement A: Ethana decolorizes broming water whereas others does not				1
	Statement B: Mixture of v	water and alcohol is us	sed in radiators of vehicles in cold countries.	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.	a) Statement B is true; S	Statement A is false.		
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false.  b) Both - Statement A and Statement B - are	c) Statement A is true; S	tatement B is false.		
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are	56 Dienocable plactic plates o	hould not be used bee		1
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.				
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because:  1	•	-		
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because: 1 a) They are made of non-biodegradable material. b) They are made of biodegradable materials.	c) They are made of tox	ic materials.	_	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because: 1 a) They are made of non-biodegradable material. b) They are made of biodegradable materials.			weight.	
Statement B: Mixture of water and alcohol is used in radiators of vehicles in cold countries.  a) Statement B is true; Statement A is false. b) Both - Statement A and Statement B - are true. c) Statement A is true; Statement B is false. d) Both - Statement A and Statement B - are false.  56. Disposable plastic plates should not be used because: 1 a) They are made of non-biodegradable material. b) They are made of biodegradable materials. c) They are made of toxic materials. d) They are made of materials with light				

57. The driving force of an ecosy	stem is:		1
a) Biomass		b) Producers	
c) Solar energy		d) Carbohydrates in plants	
58. Which group(s) of organisms	is/are not a const	ituent of a food chain?	1
A. Grass, lion, rabbit, wolf	onnor		
B. Plankton, man, fish, grassh C. Wolf, grass, snake, tiger	opper		
D. Frog, snake, eagle, grass, g	rasshopper		
a) B and D	••	b) All of these	
c) B and C		d) A and C	
59. What will happen if deer is n	issing in the give	n food chain?, Grass $ ightarrow$ Deer $ ightarrow$ Tiger.	1
a) Tiger will start eating gr	ass.	b) The population of grass will decrease.	
c) The population of tiger		d) The population of tiger will increase.	
and the population of gra		iaaaa liha.	1
60. Exposure to ultraviolet radia	ition causes eye d		1
a) Conjunctivitis c) Short-sightedness		b) Cataract d) Colour blindness	
61. Function of an ecosystem in	volves:	u) Colour billiuliess	1
a) Energy flow and nutrie		b) Energy flow only	-
c) Nutrient flow only	nt movement	d) None of the above	
62. Montreal protocol became e	ffective in:	-,	1
a) 1985		b) 1987	
c) 1992		d) 1989	
63. Match the following with the	correct response	:	1
(1) Tree	(A) Producer		
(2) Grasshopper	(B) Secondary	/ consumer	
(3) Frog	(C) Herbivore	)	
(3) Frog (4) Snake	(C) Herbivore (D) Secondary		
(4) Snake		y carnivore	
(4) Snake a) 1-D, 2-A, 3-C, 4-B	(D) Secondary	y carnivore b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C	1
(4) Snake a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D	(D) Secondary	y carnivore b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D 64. If a grasshopper is eaten by a) Primary consumer to seconsumer	(D) Secondary	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D 64. If a grasshopper is eaten by a) Primary consumer to seconsumer c) Producer to decompose	(D) Secondary a frog, then the en	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer	
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D 64. If a grasshopper is eaten by a) Primary consumer to seconsumer c) Producer to decompose	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer	
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer	
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to so consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer	
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer alts in: b) B and C	
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer alts in:	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for	(D) Secondary a frog, then the enecondary r	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D	
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to so consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia (2) Myopia	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia (2) Myopia (3) Presbyopia	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia (2) Myopia	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia (2) Myopia (3) Presbyopia (4) Astigmatism  (A) Cylindrical lens	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia (2) Myopia (3) Presbyopia (4) Astigmatism  (A) Cylindrical lens (B) Concave lens	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1
(4) Snake  a) 1-D, 2-A, 3-C, 4-B c) 1-A, 2-C, 3-B, 4-D  64. If a grasshopper is eaten by a) Primary consumer to se consumer c) Producer to decompose  65. Excessive exposure of huma A. Damage to immune system B. Damage to lungs C. Skin cancer D. Peptic ulcers a) A and C c) A and B  66. Optical fibre is used for a) All of these c) communication over lo  67. Match the following with co (1) Hypermetropia (2) Myopia (3) Presbyopia (4) Astigmatism  (A) Cylindrical lens	(D) Secondary  a frog, then the enecondary  r  ns to UV rays resum	b) 1-C, 2-B, 3-D, 4-A d) 1-B, 2-D, 3-A, 4-C nergy transfer will be from: b) Secondary consumer to primary consumer d) Producer to primary consumer nlts in: b) B and C d) A and D b) biomedical engineering	1

a) 1-C, 2-B, 3-D, 4-A c) 1-D, 2-A, 3-C, 4-B 68. Statement A : Improper functioning of rod shap length of the human eye can be increased or de	b) 1-A, 2-C, 3-B, 4-D d) 1-B, 2-D, 3-A, 4-C ped cells causes colour blindness , Statement B : The focal ecreased	1
a) Statement A is true, B is false	b) Neither statement A nor statement B is	
c) Both Statement A and B are true 69. Match the following with correct response.	true d) Statement B is true, A is false	1
<ul><li>(1) Cataract</li><li>(2) Myopia</li><li>(3) Hyper metropia</li><li>(4) Presbyopia</li></ul>		
<ul> <li>(A) Old age person unable to see near objects cle</li> <li>(B) A person can see near objects but not able to</li> <li>(C) Opacity of the lens</li> <li>(D) A person can see far objects but not able to see</li> </ul>	see for objects clearly	
a) 1-B, 2-D, 3-A, 4-C	b) 1-C, 2-B, 3-D, 4-A	
c) 1-A, 2-C, 3-B, 4-D	d) 1-D, 2-A, 3-C, 4-B	
70. When do we say a person is colour blind?		1
<ul> <li>a) When person cannot see in the light</li> <li>c) When person cannot differentiate</li> <li>between colours</li> </ul>	b) All of these d) When person cannot see in the dark	
	to its component colours. This phenomenon is called.	1
a) Reflection	b) Spectrum	
c) Dispersion	d) Refraction	
72. What is the observed colour of sky as seen from	the moon surface?	1
a) Black	b) Blue	
c) Red	d) None of these.	4
73. The lateral displacement of an incident ray pass		1
<ul> <li>a) independent of the thickness of the glass slab</li> </ul>	b) None of these	
c) is directly proportional to the thickness of the glass slab	d) inversely proportional to the thickness of the glass slab	
74. The critical angle for diamond is		1
a) 90 <sup>0</sup> c) 50 <sup>0</sup>	b) 180 <sup>0</sup> d) 24 <sup>0</sup>	
75. Danger signals are red in colour because	0) 24	1
a) red colour is least scattered     c) wavelength of red colour is less than that     of other colour	b) red colours is most scattered d) red colour looks attractive	-
76. A boy uses spectacles of focal length – 50cm. He	nce the defect of vision, he is suffering from-	1
a) Far-sightedness	b) Myopia	
c) Hypermetropia	d) Presbyopia.	
<ul><li>77. Match the following with correct response.</li><li>(1) Atmospheric refraction</li><li>(2) Scattering of light</li><li>(3) Dispersion</li></ul>		1
(4) Tyndall effect		
(A) Twinkling of star		
(B) Rainbow (C) Red colour of rising sun		
(D) White colour of clouds		
	8	

a) 1-A, 2-C, 3-B, 4-D c) 1-C, 2-B, 3-D, 4-A 78. Match the following with correct response. (1) Pupil (2) Choroid	b) 1-B, 2-D, 3-A, 4-C d) 1-D, 2-A, 3-C, 4-B	1
(3) Retina (4) Sclerotic		
<ul><li>(A) Iris, Pupil</li><li>(B) Cornea</li><li>(C) Ciliary muscles, lens</li><li>(D) Acts as a variable aperture in eye</li></ul>		
a) 1-A, 2-C, 3-B, 4-D c) 1-B, 2-D, 3-A, 4-C 79. Ability of the eye lens to adjust its focal length is	b) 1-C, 2-B, 3-D, 4-A d) 1-D, 2-A, 3-C, 4-B called	1
a) Accommodation     c) Power  80. Which of following phenomena is based on atmo		1
A. Sun appears to rise 2 minutes before and 2 mi B. Stars seen higher than they actually are C. Rainbow D. Blue colour of clear sky	nutes later	
a) A and C c) A and D	b) A and B d) B and C	
Soci	cial Science	
81. The Vernacular Press Act of 1878 was modelled	on the	1
a) American laws	b) Irish laws	
c) Australian laws	d) German laws	4
82. Study the picture and answer the question that  Which of the following option is true regarding t		1
a) The first book he printed was the	b) The first book he printed was the	
Panchtantra c) The first book he printed was the Bible.	Accordion Book d) The first book he printed was the Diamond Sutra	
83. Gutenberg printed the text in which colour?		1
a) White	b) Red	
c) Blue 84. The dust cover or the book jacket is an innova	d) Black	1
a) 20th century	b) 16th century	1
c) 18th century	d) 19th century	
	merous religious texts in vernaculars, who founded this	1
a) Ratan Naval Kishore	b) Dinesh Naval Kishore	
c) Raj Naval Kishore	d) Munshi Naval Kishore	
	9	

86. Richard M. Hoe belonged to which of the follow	wing country?	1
a) USA	b) England	
c) China	d) Brazil	
87. The press regulations of 1823 which were repe		1
a) Adams	b) Canning	
c) Benetick	d) Warren Hastings	1
88. Who wrote 'My childhood and My university	l) Candana	1
a) George Eliot c) Jane Austen	b) Cardona d) Maxim Gorky	
	overnment after publishing which of the following poems?	1
a) Resistance	b) Bande Matram	
c) Maratha	d) Shivaji's Utterences	4
90. Where was the world's first newspaper publishe		1
a) India c) Rome	b) paris d) China	
91. Sebastien Mercier was a/an		1
a) Comedian	b) Editor	
c) Novelist	d) Artist	
92. Penny chapbooks were first printed from which	of the following country ?	1
a) England	b) Canada	
c) India	d) China	
93. Who were Bronte sisters?		1
a) They were novelists	b) They were philosophers	
c) They were historians 94. Shamsul Akhbar was published in which langua	d) They were Journalists	1
a) Sanskrit	b) Persian	•
c) Urdu	d) Arabic	
95. Study the picture and answer the question that i	follows	1
95. Study the picture and answer the question that i	ionows.	1
學事。 本本 本本 本本 本本 本本 本 本 本 本 本 本	ng the picture?	
a) It is a page from the oldest Japanese book	b) It is a page from the Ninety Five Theses.	
<ul> <li>Diamond Sutra.</li> <li>c) It is a page from the Panchtantra book</li> </ul>	d) It is a page from the traditional Chinese	
OC Chimm buothers which adouble to fit of B	'accordion book'	4
96. Grimm brothers published which of the following		1
a) Novels c) Ballads	b) Folk tales d) Newspapers	
97. Who among the following is known as liberator		1
a) Dalhousie	b) Mayo	
c) Hastings	d) Charles Metcafe	
98. Press came to be made out of metal in		1
a) Eighteenth century c) Seventeenth century	b) Fifteenth century d) Sixteenth century	
	10	

99. Which among the following is an autobiography	of Rashundari Devi	1
a) Amar Jawan c) AmarJyoti	b) Amar zindagi d) Amar Jiban	
100. 'Chhote Aur Bade Ka Sawal' was written by		1
a) Tilak	b) MK Gandhi	
c) Kashibaba	d) Ambedkar	
101. The was the world's first mass-produced of		1
a) H Model Ford c) O Model Ford	b) M Model Ford d) T Model Ford	
102. Give one word for : They link national currencies	.,	1
a) Floating rate	b) Flexible rate	-
c) Exchange rate	d) Fixed rate	
103. Abolition of meant that food could be improduced in Britain.	ported at much cheaper rate than at what it could be	1
a) Corn Laws	b) Zamindari System	
c) Permanent Settlement system	d) Land to the Tiller Act	_
104. What method was used by the Portuguese and to		1
a) Guns and firearms	b) Germs and Virus d) Firepower and bombs	
c) Army and Military 105. Indentured Indian labourers were often referre	-	1
a) Poors	b) Dalits	-
c) Coolies	d) Porters	
106. Which Conference was held in July 1944 at Brett	ton Woods?	1
a) Earth Conference	b) United Nations Monetary and Financial Conference	
c) United nation Development programme	d) United Nations Financial and Monetary	
Conference	Conference	
107. In which century trade flourished and markets	expanded?	1
a) In the late sixth century	b) In the late nineteenth century	
c) In the late eighteenth century	d) In the late seventeenth century	_
108. The pre-modern world shrank greatly in which		1
a) Sixteenth c) Nineteenth	b) Tenth d) Sixth	
109. Economists has identify three types of flows wit		1
Which of the given is not a part of that flow?		_
a) The flow of trade	b) The movement of capital	
c) The flow of labour	d) The flow of technology	
110. Which of the following country was not exporting		1
a) Russia	b) China	
c) America	d) Australia	_
111. The US no longer commanded confidence a		1
a) Euro c) Peso	b) Dollar d) Pound	
	tant influencing factor that lead to the recovery after the	1
i. The emergence of the US as the dominant eco ii. Transformation of the Soviet Union from an a iii. The dominance of the Soviet Union		
a) i only	b) Only ii	
c) i, ii and iii 113. Tariff is tax imposed on a country's from t	d) ii and iii the rest of the world.	1
a) Per Capita Income	b) Imports	
c) Exports	d) National Income	
	11	

114. Income from the Indian market was utilised b India.	y Britain to pay '' for its officials who were posted in	1
a) War equipment	b) Home charges	
c) Loans	d) Import duties	
<ol> <li>John Maynard Keynes-the famous economist t promoted global economic recovery.</li> </ol>	hought that India during the Great Depression of 1929	1
a) Gold exports	b) Spices imports	
c) Opium exports	d) B. Gold imports	
116 were not equipped to cope with the chalcolonies.	lenge of poverty and lack of development in the former	1
i. International Bank for Reconstruction and E ii. Consumer Welfare Fund iii. International Monetary Fund	evelopment	
Choose the correct option.		
a) i and iii	b) i, ii and iii	
c) i and ii	d) ii only	
117. What are MNCs?		1
a) Mega National Companies	b) Medium National Corporation	
c) Multi National Corporations	d) Multi Number Companies	
118. An industrial society based on cannot	be sustained without mass consumption.	1
a) Total production	b) Production by masses	
c) Mass production	d) Gross production	
119. The peasants of Ireland became dependent on		1
a) Potato	b) Indigo	
c) Coffee	d) Tea	
120 and other asian countries became attract		1
a) America	b) China	
c) Russia	d) Australia	

## **Solution**

## **Class 10 - Mathematics**

# MULTIPLE CHOICE QUESTION EXAMINATION

## **Section A**

1. (a)

44 cm

## Explanation:

Since tangents from an external point to a circle are equal in length.

$$\therefore$$
 AS = AP = 6 cm and AB = 6 + 7 = 13 cm

$$PB = BQ = 7 \text{ cm} \text{ and } BC = 7 + 5 = 12 \text{ cm}$$

$$CQ = CR = 5 \text{ cm} \text{ and } CD = 5 + 4 = 9 \text{ cm}$$

$$RD = SD = 4 \text{ cm} \text{ and } AD = 4 + 6 = 10 \text{ cm}$$

Therefore, perimeter of quadrilateral ABCD = 13 + 12 + 9 + 10 = 44 cm

2. (d)

5 cm

## **Explanation:**

:Let point of contact of RS be A,

point of contact of QR be B,

point of contact of PQ be C

and point of contact of PS be D.

Also let 
$$AS = x$$
 and  $AR = y$ 

Now, AS = SD = x [Tangents from an external point]

$$\Rightarrow$$
PD = 4.2 - x

But PD = PC = 
$$4.2 - x$$

And QC = 
$$6.5 - PC = 6.5 - 4.2 + x = 2.3 + x$$
.....(i)

Now, again, AR = BR = y [Tangents from an external point]

$$\Rightarrow$$
QB = 7.3 - y

But QB = QC [Tangents from an external point]

$$\therefore$$
 7.3 - y = 2.3 + x  $\Rightarrow$  x + y = 5

$$\Rightarrow$$
AS + AR = 5 cm

3. (d)

30°

## Explanation:

Here  $\angle$ QPT = 60v [Angles opposite to equal sides]

And 
$$\angle PTQ = 180^{\circ} - (60^{\circ} + 60^{\circ}) = 60^{\circ}$$
 [Angle sum property of a triangle]

$$\therefore$$
 POQ = 180° - 60° = 120°

Let  $\angle OPQ = \angle OQP = x$  [Angles opposite to equal sides (Radii)]

∴ In triangle OPQ,

$$\angle POQ + x + x = 180^{\circ}$$

$$\Rightarrow$$
2x = 60°

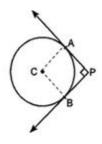
$$\Rightarrow$$
x = 30°

Therefore,  $\angle$ OPQ = 30°

4. (b)

5 cm

Explanation:



Construction: Joined OA and OB.

Here OA⊥ AP

and OB⊥BP

and PA  $\perp$ PB

Also AP = PB

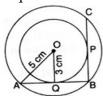
Therefore, APBO is a square.

$$\Rightarrow$$
 AP = OA = OB = 5 cm

### 5. (b)

8 cm

## **Explanation:**



Construction: Joined OP.

In right angled triangle AOQ,

$$AQ = \sqrt{(5)^2 - (3)^2} = \sqrt{25 - 9} = \sqrt{16} = 4 \text{ cm}$$

Since perpendicular from centre bisect opposite sides.

Also QB = PB = 4 cm [Tangents to a circle]

And PB = PC =  $4 \text{ cm} [OP \perp BC]$ 

$$\therefore$$
 BC = PB + PC = 4 + 4 = 8 cm

## 6. (a)

AC = BC

### **Explanation:**

Since Tangents from an external point to a circle are equal.

$$\therefore$$
 PB = BR .....(i)

Adding eq. (i) and (iii), we get

$$PB + CQ = BR + CR$$

$$\Rightarrow$$
 AP + CQ = BC [Given: PB = AP]

$$\Rightarrow$$
AQ + CQ = BC [From eq. (ii) AP = AQ]

$$\Rightarrow$$
 AC = BC

7. (d)

PQ

### **Explanation:**

PD + QB = PA + QA [Tangents from an external point to a circle are equal]  $\Rightarrow$  PD + QB = PQ

8. (c)

8 cm

In right angled triangle COE,

$$OC^2 = OE^2 + CE^2$$

$$\Rightarrow$$
(5)<sup>2</sup> = (3)<sup>2</sup> + CE<sup>2</sup>  $\Rightarrow$ CE<sup>2</sup> = 25 - 9 = 16

$$\Rightarrow$$
CE = 4 cm

Since, perpendicular from centre of a circle to the chord bisects the chord.

$$\therefore$$
 CD = CE + ED = 4 + 4 = 8 cm

9. (d)

15 cm.

Explanation:



Construction: Joined OP and OS. Draw OQ  $\perp$  PS

Here PR = PQ = 7.5 cm [Tangents to a circle from an external point]

Now, in triangles OPQ and OSQ,

 $\angle$ PQO =  $\angle$ SQO = 90° [Line segment is perpendicular from centre to point of contact]

OQ = OQ [Common]

OP = OS [Radii]

$$\therefore \Delta \text{ OPQ} \cong \Delta \text{ OSQ [SAS congruency]}$$

Therefore, 
$$PS = PQ + QS$$

$$= 7.5 + 7.5 = 15$$
 cm

10. (c)

50°

**Explanation:** 

Here  $\angle$ OAP = 90°

And  $\angle OPA = \frac{1}{2} \angle BPA$  [Centre lies on the bisector of the angle between the two tangents]

$$\Rightarrow \angle \text{OPA} = \frac{1}{2} \times 80^{\circ} = 40^{\circ}$$

Now, in triangle OPA,

$$\angle OAP + \angle OPA + \angle POA = 180^{\circ}$$

$$\Rightarrow$$
90° + 40° +  $\angle$ POA = 180°

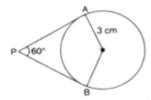
$$\Rightarrow \angle POA = 50^{\circ}$$

11. (a)

$$3\sqrt{3}$$

Explanation:

Let O be the centre. Construction: Joined OP.



Since OP bisects  $\angle$ P, therefore,  $\angle APO = \angle OPB = 30^{\circ}$  And  $\angle OAP = 90^{\circ}$ 

$$\therefore \tan 30^\circ = \frac{\mathrm{OA}}{\mathrm{AP}} \Rightarrow \frac{1}{\sqrt{3}} = \frac{3}{\mathrm{AP}}$$

 $\Rightarrow$ AP =  $3\sqrt{3}$  cm Since each tangent from an external point to a circle are equal.

Therefore, PA = PB =  $3\sqrt{3}$  cm

70°

## **Explanation:**

Here 
$$\angle OQS = \angle OQL - \angle SQL = 90^{\circ} - 50^{\circ} = 40^{\circ}$$

And 
$$\angle$$
ORS =  $\angle$ ORM -  $\angle$ SQM 90° - 60° = 30° Since OS = OQ [Radii]

$$\Rightarrow$$
  $\angle$  OSQ =  $\angle$  OQS = 40° [Angles opposite to equal sides] Again, since OS = OR [Radii]

$$\Rightarrow$$
  $\angle$ OSR =  $\angle$ ORS = 30° [Angles opposite to equal sides]

$$\therefore$$
  $\angle$ QSR =  $\angle$ OSQ+ $\angle$ OSR = 40° + 30° = 70°

13. (b)

56 cm

## Explanation:

We know that, PQ = 
$$\frac{1}{2}$$
(Perimeter of  $\triangle$ PLM)

$$\Rightarrow$$
 28 =  $\frac{1}{2}$  (Perimeter of  $\triangle$ PLM)

$$\Rightarrow$$
(Perimeter of  $\triangle$ PLM) = 28  $\times$  2 = 56 cm

100°

## **Explanation:**

Since OT bisects the  $\angle T$ .

and 
$$\angle ATB = 40^{\circ} + 40^{\circ} = 80^{\circ}$$

$$\therefore$$
 ∠AOB = 180° - 80° = 100°

## 15. (a)

4 cm

## **Explanation:**

Here  $\angle Q = 90^{\circ}$  [Angle between tangent and radius through the point of contact]

Now, in right angled triangle OPQ,

$$OP^2 = OQ^2 + PQ^2$$

$$\Rightarrow$$
(5)<sup>2</sup> = (3)<sup>2</sup> + PQ<sup>2</sup>

$$\Rightarrow$$
PQ<sup>2</sup> = 25 - 9 = 16

$$\Rightarrow$$
 PQ = 4 cm

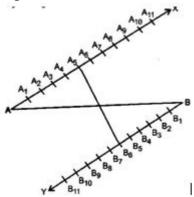
But PQ = PR [Tangents from one point to a circle are equal]

Therefore, PR = 4 cm

#### 16. (a)

 $A_5$  and  $B_6$ 

### **Explanation:**



According to the question, the point joined are  $A_5$  and  $B_6$ . The point where  $A_5B_6$  intersects the given line is

the required point.

17. (b)

7

## Explanation:

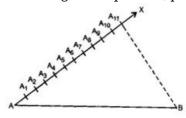
When numerator is greater than the denominator, the number of arcs should be drawn larger of m and n. Therefore, according to the question, the minimum number of points to be located at equal distances on ray BX is 7.

18. (d)

 $A_{11}$ 

## **Explanation:**

According to the question, point B is joined to A11.



19. (b)

 $135^{o}$ 

## **Explanation:**

According to the question, the angle between radii is  $180^{\circ}-45^{\circ}=135^{\circ}$ 

20. (b)

 $L_7$ 

## **Explanation:**

To divide a line segment AB in the ratio m: n, a ray AX is drawn, such that ∠BAX is an acute angle and the points A1, A2, A3, ......, Am, ....... An are located at equal distances on the ray AX and then the point B is joined to An.

Therefore, according to the question, the point M is joined to L7M.

21.

$$(\frac{10}{3}, \frac{7}{3})$$

## **Explanation:**

Let us suppose,P(-6,7),Q(x,y) is divided by R(-2,3) in the ratio of 3:4

$$(\frac{-6\times 4+3\times x}{3+4})=-2$$
 Or, $-24+3x=-14$  Or, $3x=10$ 

$$0r, x = \frac{10}{3}$$

Or, 
$$x = \frac{10}{3}$$
  
 $(\frac{7 \times 4 + 3 \times y}{3 + 4}) = 3$ 

$$0r, 28 + 3y = 21$$

Or,
$$3y = 7$$

Or,
$$y = \frac{7}{3}$$

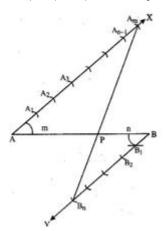
Then the co ordinate of  $Q(\frac{10}{3}, \frac{7}{3})$ 

22. (d)

 $A_5$  to  $B_7$ 

**Explanation:** 

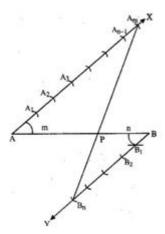
According to the question, the points to be joined are  $A_5$  to  $B_7$  because if we have to divide a line segment AB in the ratio m:n, then we draw rays AX and BY and mark the points  $A_1, A_2, \ldots, A_m$  and  $B_1, B_2, \ldots, B_n$  on rays AX and BY respectively. Then we join the point  $A_m$  to  $B_n$ 



23. (b)  $A_6$  and  $B_5$ 

## Explanation:

According to the question, the points joined are  $A_6$  to  $B_5$ . Because if we have to divide a line segment AB in the ratio m: n, then we draw rays AX and BY and mark the points  $A_1, A_2, \ldots, A_m$  and  $B_1, B_2, \ldots, B_n$  on rays AX and BY respectively. Then we join the point  $A_m$  to  $B_n$ .



24. (b) p + q

## **Explanation:**

According to the question, the minimum number of those points which are to be marked should be (Numerator + Denominator) i.e., p+q

25. (c) 1/4

Explanation:

In the figure,  $AA_1 = A_1A_2 = A_2A_3 = A_3C$ 

=1/4

26. (c)

secant

## **Explanation:**

A secant line, also simply called a secant, is a line meet two points in a circle.

27. (c)

## Aryabhatta

**Explanation:** 

Aryabhata said that circumference of a circle with a diameter of 20000 is (4+100)  $\times$ 8 +62000= 62832. And we know that the value of  $\pi$  is the ratio of the circumference to the diameter, so in this case  $\frac{62832}{20000}$ , which is incredibly 3.1416

28. (c) 
$$\sqrt{\pi}:2$$

**Explanation:** 

Let the radius of the circle be r and the side of the square be a.

Then according to the question,  $\pi r^2 = a^2 \Rightarrow a = r\sqrt{\pi}$  .....(i)

Now, Ratio of their perimeters =  $\frac{2\pi r}{4a}$ 

$$\Rightarrow$$
 Ratio of their perimeters =  $\frac{\pi r}{2a} = \frac{\pi r}{2r\sqrt{\pi}} = \frac{\sqrt{\pi}}{2}$ 

Ratio of their perimeters =  $\sqrt{\pi}:2$ 

56cm

Explanation:

Given: Area of circle = 2464 sq. cm

$$\Rightarrow \pi r^2 = 2464$$

$$\Rightarrow$$
  $r^2=rac{2464}{22} imes 7$  = 784

$$\Rightarrow r = 28 \ \mathrm{cm}$$

 $\therefore$  Diameter = 2× radius = 2 × 28 = 56 cm

30. (b) 
$$\pi (R^2 - r^2)$$

**Explanation:** 



The area of a ring having 'R' as outer radius and 'r' as inner radius is  $\pi {
m R}^2 - \pi r^2 = \pi \left( {
m R}^2 - r^2 
ight)$ 

31. (d) 
$$R^2 = r_1^2 + r_2^2$$

Explanation:

According to the question,

$$\pi \mathrm{R}^2 = \pi r_1^2 + \pi r_2^2 = \pi \left( r_1^2 + r_2^2 \right) \ \Rightarrow \mathrm{R}^2 = r_1^2 + r_2^2$$

19.6 cm

**Explanation:** 

Let diameter of the circle be  $d \, \mathrm{cm}$ .

$$\therefore$$
 Area =  $\pi r^2 = \pi \left(rac{d}{2}
ight)^2$   
 $\Rightarrow 301.84 = rac{22}{7} imes rac{d^2}{4}$   
 $\Rightarrow d^2 = rac{301.84 imes 4 imes 7}{22}$ 

$$\Rightarrow d^2 = 384.16$$
  
 $\Rightarrow d = 19.6 \text{ cm}$ 

$$R_1 + R_2 = R$$

Let required radius be R.

Then according to the question,

$$2\pi R_1 + 2\pi R_2 = 2\pi R$$

$$\Rightarrow 2\pi (\mathrm{R}_1 + \mathrm{R}_2) = 2\pi \mathrm{R}$$

$$\Rightarrow R_1 + R_2 = R$$

## **Explanation:**

Let the radius of the circle be r and side of the square be a. Then, according to question,

$$2\pi r=4a{\Rightarrow}a=rac{2\pi r}{4}=rac{\pi r}{2}$$

Now, ratio of their areas,

$$\frac{\pi r^2}{\left(\frac{\pi r}{2}\right)^2} = \frac{\pi r^2 \times 4}{\pi^2 r^2}$$

$$= \frac{\pi^2}{14}$$

$$\Rightarrow^{11} \pi r^2 : a^2 = 14 : 11$$

### **Explanation:**

Let the radius of the circle be r units.

Then, the perimeter of the circle =  $2\pi r$  units

The area of the circle =  $\pi r^2$  sq. units

According to the question,

$$2\pi r = \pi r^2$$

$$\implies r = 2 \, units$$

## 36. (b)

## Explanation:

Let the radius of the circle is r units.

... Area of the circle = Circumference of the circle

$$\Rightarrow \pi r^2 = 2\pi r$$

$$\Rightarrow r^2 = 2r$$

$$\Rightarrow r {\times} r \text{=} 2r$$

$$\Rightarrow r = 2$$
 units

# 37. (c)

## Explanation:

Let required radius be R.

Then according to the question,

$$2\pi \mathrm{R} = 2\pi r_1 + 2\pi r_2$$

$$=2\pi(r_1+r_2)$$

$$\Rightarrow$$
R =  $r_1 + r_2$   
 $\Rightarrow$ R =  $24 + 7 = 31$ cm

38. (b) 
$$\pi r + 2r$$

Let radius of the protractor be r  $\therefore$  Perimeter of protractor = Perimeter of semicircle + Diameter of semicircle

$$\Rightarrow$$
 Perimeter of protractor =  $\pi r + 2r$ 

Explanation:

Perimeter of circular protractor =  $\pi r + 2r$  =  $\frac{22}{7} imes 7 + 2 imes 7$ = 36 cm

40. (c) 
$$\frac{60}{6+\pi} \text{ cm}$$

Explanation:

Given: Length of arc + 
$$2 \times Radius = 20 cm$$

$$ho ext{divent. Length of arc} + 2 imes 
ho rac{ heta}{360^\circ} imes 2\pi r + 2r = 20 
ho rac{ heta 60^\circ}{360^\circ} imes 2\pi r + 2r = 20 
ho rac{\pi r}{3} + 2r = 20 
ho 
ho r \left(rac{\pi}{3} + 2
ight) = 20 
ho 
ho r \left(rac{6+\pi}{3}
ight) = 20 
ho 
ho r = rac{60}{6+\pi} ext{ cm}$$

## **Solution**

## Class 10 - Science

## **Multiple Choice Question Examination**

#### Section A

41. (c)

- CO -

## Explanation:

Butanone ( $CH_3COCH_2CH_3$ ) is the second member of ketones. It is a four carbon compound with ketone functional group (- CO -). The ketone functional group has two free valencies which are satisfied by one methyl (- $CH_3$ ) group and one ethyl (- $C_2H_5$ ) group.

42. (b)

Both the statements A and B are false

### **Explanation:**

**Valeric acid** is another name for pentanoic acid, **Soaps** are sodium or potassium salts of long chain fatty acids. When triglycerides in fat/oil react with aqueous NaOH or KOH, they are converted into **soap** and glycerol.

43. (a)

1-D, 2-A, 3-C, 4-B

### **Explanation:**

The oxyacetylene welding process uses a combination of ethyne ( $C_2H_2$ ) and oxygen gas to provide a high temperature flame. It is commonly used to permanently join mild steel. Alcohol meant for industrial purposes (fuel for spirit lamps) is made unfit for human consumption by adding small amounts (about 5%) of methanol to alcohol. The mixture is known as denatured spirit or denatured alcohol. Carbon dioxide turns lime water turns milky due to the formation of insoluble calcium carbonate. Esters have a fruity smell and are used in perfumes and cosmetics due to their characteristic odour.

44. (d)

Ethanol + Methanol (5%)

### **Explanation:**

Alcohol meant for industrial purposes is made unfit for human consumption by adding small amounts (about 5%) of methanol to ethanol. The mixture is known as **denatured spirit** or **denatured alcohol**. Addition of small amount of copper sulphate is added to impart a blue colour to denatured spirit so that it can be identified easily.

45. (c)

13 covalent bonds

## **Explanation:**

Butane C<sub>4</sub>H<sub>10</sub> has 3 C-C covalent bonds and 10 C-H covalent bonds. Thus, it has 13 covalent bonds.

46. (a)

 $\mathrm{C_3H_6}, \mathrm{C_2H_2}$ 

## Explanation:

Unsaturated hydrocarbons undergo addition reactions. Hence, saturated hydrocarbons like  $C_3H_8$  and  $CH_4$  (propane and methane) will not undergo addition reactions. Propene ( $C_3H_6$ ) and Ethyne ( $C_2H_2$ ) will undergo addition reactions.

47. (a)

#### **Fullerenes**

## **Explanation:**

Carbon can exist in three solid forms (Diamond, Graphite, Fullerenes) called allotropes.  $C_{60}$  and  $C_{70}$  are members of Fullerenes (Allotropes of carbon). Buckminsterfullerene contains cluster of 60 carbon atoms joined together to form spherical molecules.

48. (d)

Alcohol in the breath cause a chemical change which is registered by the breath analyzer machine.

## **Explanation:**

A breath-analyzer is a device for estimating blood alcohol content (BAC). Breath-analyzers do not measure blood alcohol content or concentration directly. The direct measurement requires the analysis of a blood sample. Instead, they estimate the BAC indirectly by measuring the amount of alcohol in one's breath. Alcohol in the breath causes a chemical change. This chemical change is registered by the breath-analyzer machine.

49. (d)

X

## **Explanation:**

 $C_{10}H_{21}COOH$ : Hendecanoic acid (also known as undecanoic acid, undecylenic acid and undecylic acid) is a naturally occurring carboxylic acid. It has a melting point in the range 28 - 31  $^{0}$ C. It is a low melting solid.

 $C_{19}H_{39}COOH$ : Arachidic acid or eicosanoic acid is a white crystalline solid at room temperature (25 °C). It has a melting point in the range 74 - 76 °C

50. (a)

All of these

#### **Explanation:**

**Hardness of water** is caused by magnesium and calcium salts. Calcium and magnesium dissolved in water are the two most common minerals that make water hard. Temporary hardness is a type of water hardness caused by the presence of dissolved bicarbonate minerals (calcium bicarbonate and magnesium bicarbonate).

51. (a)

Both the statements A and B are true.

## Explanation:

Oxyacetylene flame is used for welding purposes. The oxyacetylene welding process uses a combination of oxygen and acetylene ( $C_2H_2$ ) gas to provide a high temperature flame. It is commonly used to join mild steel permanently.

Ethyne ( $C_2H_2$ ) reacts with HCl in the presence of  $HgCl_2$  to from vinyl chloride or chloro ethane  $H_2C$ =CHCl. This colorless compound is an important industrial chemical. It is chiefly used to produce polyvinyl chloride (PVC).

52. **(a)** 

Decolorizing agent

#### **Explanation:**

**Activated charcoal** is activated carbon. It is a form of carbon processed to have small, low-volume pores that increases the surface area available for adsorption or chemical reactions. Activated charcoal is used as a **decolorizing agent** in the sugar industry.

53. **(c)** 

Brisk effervescence

A brisk effervescence of carbon dioxide gas will be observed when **ethanoic acid** reacts with solid **sodium bicarbonate**. This reaction is used is a test for ethanoic acid.

CH<sub>3</sub>COOH + NAHCO<sub>3</sub> --> CH<sub>3</sub>COONa + H<sub>2</sub>O + CO<sub>2</sub>

54. (d)

1-A, 2-C, 3-B, 4-D

## **Explanation:**

Hydrogenation is an addition reaction. The addition of hydrogen to an unsaturated hydrocarbon to obtain a saturated hydrocarbon is called hydrogenation. Carbon tetrachloride is a product of a substitution reaction - a characteristic property of saturated hydrocarbons. Alcohol meant for industrial purposes is made unfit for human consumption by adding small amounts (about 5%) of methanol to alcohol. The mixture is known as denatured spirit or denatured alcohol. Ethyl ethanoate is a product of esterification.

 $CH_3COOH + C_2H_5OH 
ightarrow CH_3COOC_2H_5 + H_2O$ 

55. (a)

Statement B is true; Statement A is false.

### Explanation:

Bromine water test is a test for unsaturated hydrocarbons. Ethane undergoes addition reaction and decolorizes bromine water. Similarly, ethyne also decolorizes bromine water.

Mixture of water and alcohol is used in radiators of vehicles in cold countries. Alcohol is used for antifreeze mixture. Antifreeze is an additive which lowers the freezing point of a water-based liquid.

56. (a)

They are made of non-biodegradable material.

## Explanation:

Disposable plastic plates should not be used because they are made of non-biodegradable material. Non-biodegradable substances may be inert, are not broken down and persist in the environment for a long time. They may thus cause harm to the various members of the ecosystem.

57. (c)

Solar energy

## Explanation:

The driving force of an ecosystem is solar energy. A food chain in an ecosystem always starts with photosynthesis. The autotrophs or the producers are at the first trophic level. They fix up the solar energy and make it available for heterotrophs or the consumers.

58. (c)

B and C

## **Explanation:**

Organisms in groups B and C do not make up a food chain. A food chain is a series of organisms in order of who eats whom. In B, grasshopper is out of place. In D, nobody out of wolf, snake or tiger eats grass.

59. (c)

The population of tiger will decreases and the population of grass will increase.

#### **Explanation:**

If deer is missing from the given food chain, the population of tiger will decrease and the growth of grass will increase. A missing link in a food chain will create an imbalance in the ecosystem.

60. (b)

Cataract

Exposure to ultraviolet (UV) radiation can cause eye diseases like the cataract. A cataract is a clouding of the lens in the eye which leads to a decrease in vision.

61. (a)

Energy flow and nutrient movement

#### **Explanation:**

Different materials in an ecosystem are cycled in separate biogeochemical cycles. Essential nutrients like nitrogen, carbon, oxygen and water are changed from one form to another in these biogeochemical cycles. Producers in an ecosystem fix up the solar energy and make it available for the next trophic levels.

62. (d)

1989

## Explanation:

The **Montreal Protocol** became effective in 1989. The **Montreal Protocol** is a protocol to the Vienna Convention for the Protection of the Ozone Layer. It is an international treaty designed to protect the ozone layer by phasing out the production of numerous substances that are responsible for ozone depletion.

63. (c)

1-A, 2-C, 3-B, 4-D

## Explanation:

A tree is a producer (autotroph). A grasshopper is a herbivore (primary consumer). A frog is secondary consumer.

(1) Tree	(A) Producer
(2) Grasshopper	(C) Herbivore
(3) Frog	(B) Secondary consumer
(4) Snake	(D) Secondary carnivore

#### 64. (a)

Primary consumer to secondary consumer

## **Explanation:**

If a grasshopper is eaten by a frog, then the energy transfer will be from a primary consumer to a secondary consumer. A grasshopper is a primary consumer. A frog is a secondary consumer.

65. (a)

A and C

#### **Explanation:**

Excessive exposure of humans to UV (ultraviolet) rays results in damage to the immune system. Exposure to UV radiation is also a risk factor for most skin cancers.

66. (a)

All of these

### Explanation:

Applications of optical fiber include:

- **1. Communication:** Telephone transmissoin method usese fiber-optic cables. Optical fibres transmit energy in the form of light pulses.
- **2. Medical uses:** Optical fibres are well suited for medical use. They can be made in extremely thin, flexible stands for insertion into the blood vessels, lungs, and other hollow parts of the body. Optical fibers are used in a number of instruments that enable doctors to view internal body parts without having to perform surgery.

**3. Simple uses:** The simplest application of optical fibers is the trnsmission of light to locations otherwise hard to reach.

67. (a)

1-C, 2-B, 3-D, 4-A

## **Explanation:**

For correction of hypermetropic eye or long-sighted eye, a convex lens is used.

For correction of myopic eye or short-sighted eye, a concave lens is used.

For correction of **presbyopia**, a bifocal lens (upper position consisting of concave lens and lower portion consisting of convex lens) is used.

Astigmatism can be corrected by using cylindrical lenses.

68. (d)

Statement B is true, A is false

## **Explanation:**

Colour blindness is said to occur when a person cannot differentiate between colours although his vision may otherwise be normal. Its causes are genetic and till date it has no cure. Colour blindness has no connection with improper functioning of rod shaped cells.

Ciliary muscle helps to change the curvature of eye lens and hence increases or decreases its focal length so that we can see the object clearly, placed at different position.

69. (b)

1-C, 2-B, 3-D, 4-A

## Explanation:

Cataract: The image can not be seen distinctly because eye lens become milky and cloudy.

**Myopia** (Near Sightedness): A person can see nearby object clearly, but cannot see distant object distinctly. Image formed in front of the retina.

**Hypermetropia** (Far-Sightedness): A person cannot see nearby object clearly, but can see distant object distinctly. Image formed at a point behind the retina.

**Presbyopia**: As we become old, the power of accommodation of the eye usually decreases (due to weakening of cilliary muscles), the near point gradually recedes away. This defect is called Presbyopia. Person may suffer from both myopia and hypermetropia.

70. (c)

When person cannot differentiate between colours

## **Explanation:**

Colour blindness is said to occur when a person cannot differentiate between colours although his vision may otherwise be normal. Causes of colour blindness are genetic and till date it has no cure.

71. (c)

Dispersion

## Explanation:

Dispersion is the phenomena of splitting of white light into its constituent seven colours (VIBGYOR) on passing through a glass prism.

72. (a)

Black

## Explanation:

The sky appears blue due to scattering of the blue colour by the earth's atmosphere. In moon there is no atmosphere hence nothing to scatter light. Thus, the sky appears dark as seen from the moon surface.

73. (c)

is directly proportional to the thickness of the glass slab

The lateral displacement of an incident ray passing out of a rectangular glass slab is directly proportional to the thickness of glass slab, angle of incidence, and refractive index however it is inversely proportional to the wavelength of incident light.

74. (d)

 $24^{0}$ 

## **Explanation:**

The critical angle for diamond is equal to 24.4° (approx. 24<sup>0</sup>), so that once light gets into diamond, it is very likely to be totally reflected internally. By cutting the diamond suitably, multiple internal reflections can be made to occur.

As, the sine of the critical angle is equal to the reciprocal of the refractive index of that material i.e.

$$\sin c = \frac{1}{\mu} \text{ or } c = \sin^{-1} \left(\frac{1}{\mu}\right)$$
 $c = \sin^{-1} \left(\frac{1}{2.42}\right) [\text{refractive index of diamond} = 2.42]$ 
 $c = \sin^{-1} (0.413)$ 
 $c = 24.4^{\circ}$ 

75. (a)

red colour is least scattered

#### **Explanation:**

Red colour scattered the least when strikes by the small particle of fog and smoke because it has the maximum wavelength (visible spectrum). Hence at large distance also, we can see the red colour clearly.

76. (b)

Myopia

### **Explanation:**

By convention, focal length of concave lens is taken as negative. For correction of myopic eye or near-sighted eye, a concave lens is used.

77. (a)

1-A, 2-C, 3-B, 4-D

### **Explanation:**

**Twinkling of stars** is due to atmospheric refraction.

The **rainbow** is a formed due to the dispersion and total internal reflection of sunlight by the tiny water droplet, present in atmosphere.

The **rising sun appears red** because the sun is near horizon, and therefore the sunlight has to travel larger distance in atmosphere which scatters away most of the blue light (shorter wavelength) by the atmospheric particles.

The phenomenon of scattering of light by the colloidal particles is known as Tyndall Effect. Light of all wave lengths are scattered equally by the clouds and hence **clouds appears white**.

78. (d)

1-D, 2-A, 3-C, 4-B

#### **Explanation:**

**Pupil** is the black opening (**variable aperture**) which regulates and controls the amount of light entering the eve.

**Choroid** is the middle layer. It is modified at the front to form the **Iris**. Iris contains radial and circular muscles. Iris regulates the size of the **pupil**.

**Retina** is the innermost layer. **Ciliary muscles** helps to change the curvature of **eye lens** and hence changes its focal length so that a sharp image is formed on the retina of an object placed at different

position.

**Scleroid** is the outermost part of the eye. The front transparent part through which the light enters the eye is called **cornea**.

79. (a)

Accommodation

## **Explanation:**

The ability of the eye to focus both near and distant objects, by adjusting the focal length of the eye lens, is called the accommodation of the eye.

80. (b)

A and B

## Explanation:

Both of these phenomena are due to atmospheric refraction. The temperature and density of different layer of atmosphere keeps varying. When the light enters the earth's atmosphere it undergoes refraction continuously, due to changing refractive index i.e. light travels from rarer to denser medium and hence It bends towards the normal successively.

## Solution

## Class 10 - Social Science

## MULTIPLE CHOICE QUESTION EXAMINATION

#### **Section A**

#### 81. (b) Irish laws

## **Explanation:**

In 1878'vernacular press act was passed, modelled on the irish press laws. It provided the govt with extensive rights to censor reports and editorials in the vernacular press.

#### 82. (c)

The first book he printed was the Bible.

## **Explanation:**

It is a Portrait of Johann Gutenberg. The first book he printed was the Bible.

### 83. (d) Black

### **Explanation:**

Gutenberg printed the text in black, leaving spaces where the colour could be filled in later.

## 84. (a) 20th century

### **Explanation:**

The dust cover or the book jacket is the detachable outer cover, usually made of paper and printed with text and illustrations. It is an innovation of 20th century.

## 85. (d) Munshi Naval Kishore

#### **Explanation:**

The Naval Kishore Press (NKP) was founded in Lucknow in 1858 by Munshi Naval Kishore and grew in the following decades to one of India's most important publishing houses.

### 86. (a) USA

## **Explanation:**

Richard M. Hoe, the inventor of power driven cylinderical press was the citizen of USA. The main feature of his press was that it was capable of printing 8000 sheets per hour.

## 87. (a) Adams

### **Explanation:**

Adams imposed restrictions on the press in the year 1923. The regulations of 1923 were more stringent than any that had been in force earlier. These restrictions were chiefly against Indian language newspapers.

### 88. (d) Maxim Gorky

## **Explanation:**

My childhood and My university was written by Maxim Gorky. This book provide glimpses of the struggles of the poor people against grim obstacles.

## 89. (d) Shivaji's Utterences

### **Explanation:**

Bal Gangadhar Tilak wrote a poem namely Shivaji's Utterences. This poem was published by Tilak in Kesari and for this he was arrested by the British Government.

#### 90. (c) Rome

## **Explanation:**

The first newspaper was published in Rome nemely Acta Diurna.It was published around 59 BC.

## 91. (c) Novelist

## **Explanation:**

Mercier was a novelist who declared that printing press is the most powerful engine of progress.

## 92. (a) England

**Explanation:** 

Penny Chapbooks were printed from England. These were carried by petty pedlars known as chapmen and sold for a penny so that even the poor could buy them.

## 93. (a) They were novelists

**Explanation:** 

The best known novelists in the nineteenth century were Brounte sisters. Their writings became important in defining a new type of women.

### 94. (b) Persian

Explanation:

Shamsul Akhbar, a persian newspaper was published from the year 1822 along with another persian newspaper Jam-i-Jahan Nama.

#### 95. (a)

It is a page from the oldest Japanese book - Diamond Sutra.

## **Explanation:**

The oldest Japanese book, printed in AD 868, is the Buddhist Diamond Sutra, containing six sheets of text and woodcut illustrations.

#### 96. (b) Folk tales

**Explanation:** 

About two-hundred years ago, Grimm brothers published a collection of folk tales. The Grimms didn't write these stories, they collected tales that had been handed down from generation to generation.

### 97. (d) Charles Metcafe

**Explanation:** 

Metcafe repealed the 1823 regulations with the press act of 1835 that is why he earned the epithet of liberator of the Indian press.

## 98. (a) Eighteenth century

Explanation:

By the late eighteenth century, the press came to be made out of metal. Through the nineteenth century, there were a series of further innovations in printing technology.

### 99. (d) Amar Jiban

**Explanation:** 

Amar Jiban, published in 1876, is the name of Rashundari Devi's autobiography and is the first autobiography written by an Indian woman.

## 100. (c)

Kashibaba

## **Explanation:**

Kashibaba, a millworker wrote and published 'Chhote Aur Bade Ka Sawal' in the year 1938 to show the links between caste and class exploitation.

## 101. (d)

T Model Ford

## **Explanation:**

The T Model Ford was the world's first mass-produced car.

## 102. (c) Exchange rate

**Explanation:** 

Exchange rates: They link national currencies for purposes of international trade.

## 103. (a) Corn Laws

**Explanation:** 

Abolition of Corn Laws meant that food could be imported at much cheaper rate than at what it could be produced in Britain. British farm produce was unable to compete with cheaper imports.

#### 104. (b) Germs and Virus

**Explanation:** 

The germs were used as a powerful weapon by the Portuguese and the Spanish for the colonisation of America. When the Europeans reached there, they carried the germs of small pox along with them. The disease wiped off the whole communities in certain parts of America. And thus, the Europeans could easily get control of the Americas.

### 105. (c) Coolies

**Explanation:** 

Coolie: Unskilled 'native' labourers. Indentured Indian labourers were often referred to as 'coolies' in Trinidad.

## 106. (b) United Nations Monetary and Financial Conference

**Explanation:** 

United Nations Monetary and Financial Conference was held in July 1944 at Bretton Woods in New Hampshire, USA.

## 107. (b) In the late nineteenth century

Explanation:

Trade flourished and markets expanded in the late nineteenth century. But this was not only a period of expanding trade and increased prosperity. It is important to realise that there was a darker side to this process.

### 108. (a) Sixteenth

**Explanation:** 

The pre-modern world shrank greatly in the sixteenth century after European sailors found a sea route to Asia and also successfully crossed the western ocean to America

#### 109. (d)

The flow of technology

#### **Explanation:**

- i. The first is the flow of trade which in the nineteenth century referred largely to trade in goods.
- ii. The second is the flow of labour the migration of people in search of employment.
- iii. The third is the movement of capital for short-term or long-term investments over long distances.

### 110. (b) China

**Explanation:** 

Countries which were exporting food grain to Britain:Russia,America and Australia

## 111. (b) Dollar

**Explanation:** 

The US dollar now no longer commanded confidence as the world's principal currency. The dollar could not maintain its value in relation to gold. Thus the system of fixed exchange rate collapsed and the new system of floating exchange rate began.

#### 112. (c)

i, ii and iii

## Explanation:

The recovery after the Second World War was influenced by two important factors:

## First:

- a. The emergence of the US as the dominant economic, political and military power in the west.
- b. Transformation of the Soviet Union from an agrarian economy into a world power.

Second: The second was the dominance of the Soviet Union.

## 113. (b) Imports

#### **Explanation:**

Tariff: Tax imposed on a country's imports from the rest of the world. Tariffs are levied at the point of entry, i.e., at the border or the airport.

## 114. (b) Home charges

**Explanation:** 

Income from the Indian market was utilised by Britain to serve its other colonies and also to pay 'home charges' for its officials who were posted in India.

### 115. (a) Gold exports

**Explanation:** 

Explanation: John Maynard Keynes-the famous economist thought that India gold exports during the Great Depression of 1929 promoted global economic recover.

### 116. (a) i and iii

Explanation:

The IMF and the International Bank for Reconstruction and Development (World Bank) were designed to meet the financial needs of the industrial countries. They were not equipped to cope with the challenge of poverty and lack of development in the former colonies.

## 117. (c) Multi National Corporations

**Explanation:** 

Multinational corporations (MNCs) are large companies that operate in several countries at the same time.

## 118. (c) Mass production

**Explanation:** 

An industrial society based on mass production cannot be sustained without mass consumption.

### 119. (a) Potato

**Explanation:** 

The peasants of Ireland became so dependent on potato that when disease destroyed the potato crop in the mid-1840s, hundreds of thousands died due to starvation. This famine is known as Irish Famine.

#### 120. (b)

China

### **Explanation:**

China became attractive destinations for investment: This is because of the low-cost structure of the Chinese economy, most importantly its low wages.

Wages were relatively low in countries like China. Thus they became destinations for investment by foreign

MNCs competing to capture world markets.